



Title: Coordination Chemistry

Objectives: The major objectives of this course are: (a) To understand the key features of coordination compounds, including the variety of structures, oxidation numbers and electronic configurations, coordination numbers, ligands, chelates, bonding and stability of complexes. (b) To be able to describe the stability of metal complexes and to become familiar with some applications of coordination compounds.

Outcomes: After completing this course the students are able to:

1. Understand the coordination numbers, stability and geometric shapes of the complexes
2. Study and compare between the various theories of coordination
3. Explain the reaction mechanisms of coordination compounds

Course Code: CH-807

Credit Hours: 3-0

Course Contents:

1. Introduction to transition metals
2. Historical development of coordination compounds and Werner Theory
3. IUPAC nomenclature of complexes
4. Coordination number (CN) and geometry (shape) of complexes (CN 2-6)
5. Isomerism and classification of isomers
6. Bonding Theories
7. Crystal Field Theory (CFT)
8. Molecular Orbital Theory (MOT)
9. d-d spectrum theory
10. Kinetic and Mechanism
11. Reactions of Coordination Compounds –Stabilization of complexes
12. Application of coordination compounds

Course Contents with proposed contact Hours (Weekly plan):

Lecture wise Breakdown

W#1	Topics	Lect #
1	Introduction to Transition Metals	1
	Lab-1: Introduction to Computational Methods Software Installation – SuperComputer Accounts	2-3
2	Historical development of coordination compounds and Werner Theory Isomerism and classification of isomers	4
	Lab-2: Tutorial -1	5-6
3	IUPAC nomenclature of complexes	7
	Lab-3: Tutorial – 2 Group Discussion about the Research Project for this Course	8-9
4	Coordination number (CN) and geometry (shape) of complexes (CN 2-6)	10

	Lab-4: Tutorial – 3 Finalization of Research Projects ASSIGNMENT – 1: Term Paper – Review article (seminal paper - publication last 5 to 10 years) on (any) topic of Coordination Complexes/ Compounds (Your Choice) to be submitted in 02 phases. Phase – 1: Abstract Submission within 02 weeks, Phase –II: Review article submission within 02 months.	11-12
5	Lab -5: Model Geometries of Coordination Compounds/Complexes	13 14-15
6	Bonding Theories: a. VBT b. Hybridization Lab-6: Geometry Optimization	16 17-18
7	Crystal Field Theory Lab-7: Assignment-2: Independent work on the Geometry modeling and Optimization	19 20-21
8	Revision of all topics till date Assignment-2: Result Submission	22 23-24
9	Mid Semester Exam	25-27
10	Molecular Orbital Theory Lab-8: Frequency Calculations for Validation	28 29-30
11	d-d Spectrum Theory/ Electronic Spectra Lab-9: Single Point Energy Calculations	31 32-33
12	d-d Spectrum Theory/ Electronic Spectra Revision Submission of “ Introduction ” section on the Coordination Compounds selected for study	34 35-36
13	Kinetics and Mechanism Lab-10: Natural Bond Order Calculations Submission of “ Methodology ” section on the Coordination Compounds selected for study	37 38-39
14	Reaction and Catalysis Lab-11: Data Compilation	40 41-42
15	Reactions of Coordination Compounds – Stabilization of complexes Lab-12: Data Validation	43 44-45
16	Application of Coordination Compounds Lab-13: Project Report Submission	46 47-48
17	Presentations + Revision + Problem Solving Assignment-1: Paper Submission	49-51
18	End Term 😊	52-54

Details of lab work/workshop practice, if applicable:

Lab sessions will mainly focus on the hands-on training in connection with the lectures taught in class. Details are mention in the week wise breakdown.

Recommended reading, including textbooks, reference books with dates

1. Bridget Kent, “Advanced Inorganic Chemistry”, NY Research Press, 2019

2. Robert Crabtree, *The Organometallic Chemistry of the Transition Metals*, 7th Edition, Wiley & Sons In, **2019**.
3. D. Shriver and P. Atkins, *Inorganic Chemistry*, 5th Edition. Oxford University Press, **2010**.
4. Gary L. Miessler and Donald A Tarr, *Inorganic Chemistry*, 4th Edition, Pearson Prentice Hall, **2010**.
5. "Encyclopedia of Computational Chemistry", 5th volume, John Wiley and Sons, Inc. 1998.
6. Relevant Publications

Nature of Assessments

Homework/ Assignments:	5%
Quizzes:	5%
MSE:	30%
Final Exam:	40%
Project:	20%